

## Novel *ent*-kaurane dimers from *Isodon rubescens* var. *rubescens*

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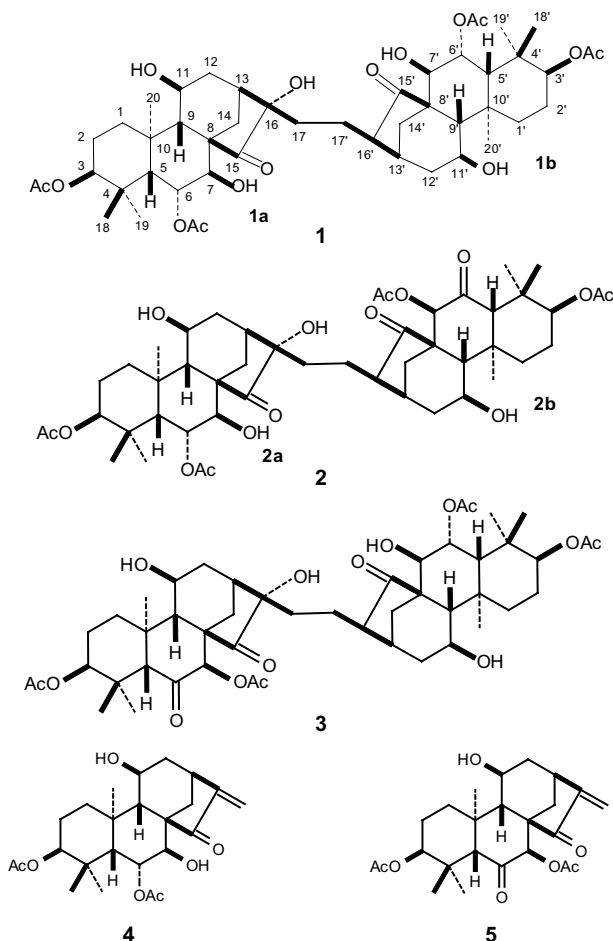
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**Abstract**—Three novel asymmetric *ent*-kaurane dimers xindongnins M–O (**1**–**3**) were isolated from *Isodon rubescens* var. *rubescens*. Their structures were elucidated by spectroscopic methods including 2D NMR analysis. The relative stereochemistry of **1** was determined by single crystal X-ray diffraction, which also confirmed the unique linkage of a single carbon–carbon bond between the two subunits of this dimer. A biogenetic pathway was proposed for the formation of these dimers.  
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In our continuing phytochemical investigation of *Isodon rubescens* complex,<sup>1a–f</sup> we examined, with the aid of analytical and preparative HPLC, the aqueous acetone extract of the leaves of *I. rubescens* var. *rubescens* collected in Shangcheng Prefecture, Henan Province of China,<sup>2</sup> as a result, three novel asymmetric *ent*-kaurane dimers were isolated, together with two known compounds melissoidesin G (**4**)<sup>3</sup> and xindongnin A (**5**).<sup>4</sup> The structures of these dimers were elucidated by the analysis of their HRMS and NMR data, especially 2D NMR spectra. These were the first examples of *ent*-kaurane dimers that possessed a rare linkage of a single carbon–carbon bond between the two structural subunits. This unique linkage was finally confirmed by single crystal X-ray analysis.

Xindongnin M (**1**) was obtained as colorless cubes. Its molecular formula was determined to be C<sub>48</sub>H<sub>70</sub>O<sub>15</sub> by positive HRFABMS (found 887.4793, calcd 887.4811 for C<sub>48</sub>H<sub>71</sub>O<sub>15</sub>).<sup>5</sup> Its <sup>13</sup>C NMR spectrum (Table 1) confirmed this formula, and exhibited carbon signals of two diterpene units (**1a** and **1b**) bearing four acetoxy groups in total. Each of the units was further indicated to have a kaurane skeleton by the characteristic signals of three methyl groups (C-18, 19, and 20), three methine carbons (C-5, 9, and 13), and three quaternary carbons



**Keywords:** *ent*-Kaurane dimer; Xindongnins M–O; *Isodon rubescens* var. *rubescens*.

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**Table 1.**  $^1\text{H}$  (500 MHz) and  $^{13}\text{C}$  NMR (125 MHz) data of compounds **1–5** in  $\text{C}_5\text{D}_5\text{N}$  ( $\delta$ ,  $J$  in ppm)

No	<b>1</b>		<b>2</b>		<b>3</b>		<b>4</b>	<b>5</b>
	$^1\text{H}$	$^{13}\text{C}$	$^1\text{H}$	$^{13}\text{C}$	$^1\text{H}$	$^{13}\text{C}$	$^{13}\text{C}$	$^{13}\text{C}$
1 $\alpha$	1.44–1.67 <sup>a</sup>	35.5 t	1.67 <sup>h</sup>	35.5 t	1.61 (m)	35.0 t	35.5 t	33.5 t
1 $\beta$	1.44–1.67 <sup>a</sup>		1.47 <sup>i</sup>		1.44 (m)			
2 $\alpha$	1.87–1.96 <sup>b</sup>	22.9 t	1.54–1.64 <sup>j</sup>	22.9 t	1.52–1.58 <sup>o</sup>	22.6 t	23.6 t	22.6 t
2 $\beta$	1.44–1.67 <sup>a</sup>		1.54–1.64 <sup>j</sup>		1.52–1.58 <sup>o</sup>			
3 $\alpha$	4.73 (br s)	78.6 d	4.73 (br s)	78.7 d	4.66 (br s)	77.1 d	78.3 d	77.2 d
4		37.1 s		37.1 s		35.8 s	37.3 s	35.8 s
5 $\beta$	2.60 (s)	42.4 d	2.60 (s)	42.4 d	3.58 (s)	55.0 d	41.9 d	54.9 d
6 $\beta$	5.60 (br s)	71.9 d	5.59 (br s)	71.9 d		206.4 s	71.4 d	202.5 s
7 $\alpha$	4.33 (br s)	74.6 d	4.31 (br s)	74.6 d	5.55 (br s)	80.9 d	72.9 d	80.6 d
8		50.5 s		50.1 s		53.1 s	49.5 s	53.5 s
9 $\beta$	2.62 (s)	60.4 d	2.60 (s)	60.4 d	2.80 (s)	59.8 d	58.6 d	59.2 d
10		38.0 s		38.2 s		44.5 s	38.1 s	44.9 s
11 $\alpha$	4.33 (br s)	64.4 d	4.31 (br s)	64.4 d	4.24 (br s)	63.9 d	65.7 d	64.7 d
12 $\alpha$	2.54 <sup>e</sup>	32.5 t	2.50–2.54 <sup>k</sup>	32.5 t	2.38 <sup>p</sup>	33.7 t	41.3 t	40.7 t
12 $\beta$	2.54 <sup>e</sup>		2.50–2.54 <sup>k</sup>		1.99 <sup>q</sup>			
13 $\alpha$	2.64 <sup>d</sup>	40.9 d	2.60 (m)	40.8 d	2.56 (m)	39.3 d	37.8 d	36.9 d
14 $\alpha$	2.56 (d, 12.0)	35.2 t	2.44 <sup>l</sup>	36.6 t	2.37 <sup>p</sup>	35.7 t		34.4 t
14 $\beta$	2.18 (dd, 4.0, 12.0)		2.16 <sup>m</sup>		2.01 <sup>q</sup>			
15		226.5 s		225.7 s		213.1 s	213.0 s	206.7 s
16		80.6 s		80.5 s		80.6 s	149.7 s	151.1 s
17a	3.17 (m)	30.7 t	3.05 (m)	30.5 t	2.94 (m)	30.5 t	114.0 t	112.7 t
17b	2.39 <sup>e</sup>		2.26 <sup>n</sup>		2.63 (m)			
18	1.02 (3H, s)	28.0 q	1.00 (3H, s)	28.0 q	1.00 (3H, s)	26.8 q	27.9 q	27.0 q
19	1.01 (3H, s)	23.5 q	1.00 (3H, s)	23.5 q	1.31 (3H, s)	21.7 q	23.5 q	22.1 q
20	1.39 (3H, s)	19.6 q	1.37 (3H, s)	19.6 q	1.06 (3H, s)	18.4 q	19.4 q	18.6 q
1 $\alpha'$	1.44–1.67 <sup>a</sup>	35.4 t	1.92 (m)	33.4 t	1.67–1.81 <sup>r</sup>	34.0 t		
1 $\beta'$	1.44–1.67 <sup>a</sup>		2.26 <sup>n</sup>		1.67–1.81 <sup>r</sup>			
2 $\alpha'$	1.87–1.96 <sup>b</sup>	22.9 t	1.54–1.64 <sup>j</sup>	22.6 t	1.52–1.58 <sup>o</sup>	22.3 t		
2 $\beta'$	1.44–1.67 <sup>a</sup>		1.54–1.64 <sup>j</sup>		1.52–1.58 <sup>o</sup>			
3 $\alpha'$	4.73 (br s)	78.6 d	4.65 (br s)	77.5 d	4.72 (br s)	78.3 d		
4'		37.1 s		36.1 s		36.8 s		
5 $\beta$	2.57 (s)	42.4 d	3.50 (s)	55.7 d	2.55 (s)	42.1 d		
6 $\beta'$	5.60 (br s)	71.9 d		206.5 s	5.51 (br s)	71.5 d		
7 $\alpha'$	3.93 (br s)	73.7 d	5.31 (br s)	80.6 d	3.92 (br s)	73.3 d		
8'		50.1 s		53.8 s		50.2 s		
9 $\beta'$	2.43 (s)	58.4 d	2.64 (s)	58.7 d	2.41 (s)	58.0 d		
10'		38.2 s		44.4 s		37.7 s		
11 $\alpha'$	4.28 (br s)	64.0 d	4.20 (br s)	63.9 d	4.26 (br s)	63.7 d		
12 $\alpha'$	2.00 (m)	34.0 t	1.79 (m)	34.2 t	2.07 (m)	35.6 t		
12 $\beta'$	2.47 (m)		1.69 <sup>h</sup>		1.23 (m)			
13 $\alpha'$	2.54 (m)	33.9 d	2.44 <sup>l</sup>	32.6 d	2.54 (m)	33.5 d		
14 $\alpha'$	2.58 (d, 12.0)	33.5 t	2.12 <sup>m</sup>	34.3 t	2.59 (d, 12.0)	30.6 t		
14 $\beta'$	1.24 (dd, 4.0, 12.0)		1.49 <sup>i</sup>		2.35 <sup>s</sup>			
15'		225.7 s		214.3 s		226.3 s		
16'	2.39 <sup>e</sup>	57.3 s	2.30 (m)	57.1 d	2.38 <sup>s</sup>	56.8 d		
17a'	2.97 (m)	20.8 t	2.89 (m)	20.5 t	2.90 (m)	20.5 t		
17b'	2.66 <sup>d</sup>		2.47 <sup>l</sup>		2.50 (m)			
18'	1.02 (s, 3H)	28.0 q	1.00 (3H, s)	27.1 q	1.00 (3H, s)	27.7 q		
19'	1.01 (s, 3H)	23.5 q	1.30 (3H, s)	22.1 q	1.01 (3H, s)	23.3 q		
20'	1.32 (s, 3H)	19.4 q	1.00 (3H, s)	18.6 q	1.31 (3H, s)	19.1 q		
Oac		170.1 s		170.1 s		170.0 s	170.6 s	169.7 s
		169.9 s		169.9 s		169.7 s	169.4 s	169.6 s
		170.1 s		169.8 s		169.6 s		
		169.9 s		169.7 s		169.5 s		
	2.10 (s, 3H) <sup>f</sup>	20.9 q	2.24 (3H, s)	21.3 q	2.26 (3H, s)	21.1 q	21.6 q	20.9 q
	1.84 (s, 3H) <sup>g</sup>	21.3 q	2.03 (3H, s)	21.1 q	2.10 (3H, s)	20.8 q	21.3 q	20.8 q
	2.10 (s, 3H) <sup>f</sup>	20.9 q	1.88 (3H, s)	20.9 q	1.90 (3H, s)	20.6 q		
	1.84 (s, 3H) <sup>g</sup>	21.3 q	1.83 (3H, s)	20.7 q	1.85 (3H, s)	20.5 q		
OH-7	6.53 (s)		6.51 (s)					
OH-11	6.46 (s)		6.41 (s)		6.49 (s)			
OH-16	7.46 (s)		7.37 (s)		7.48 (s)			
OH-7'	6.47 (s)				6.46 (s)			
OH-11'	6.27 (s)		6.32 (s)		6.24 (s)			

<sup>a–s</sup>Signals overlapped.

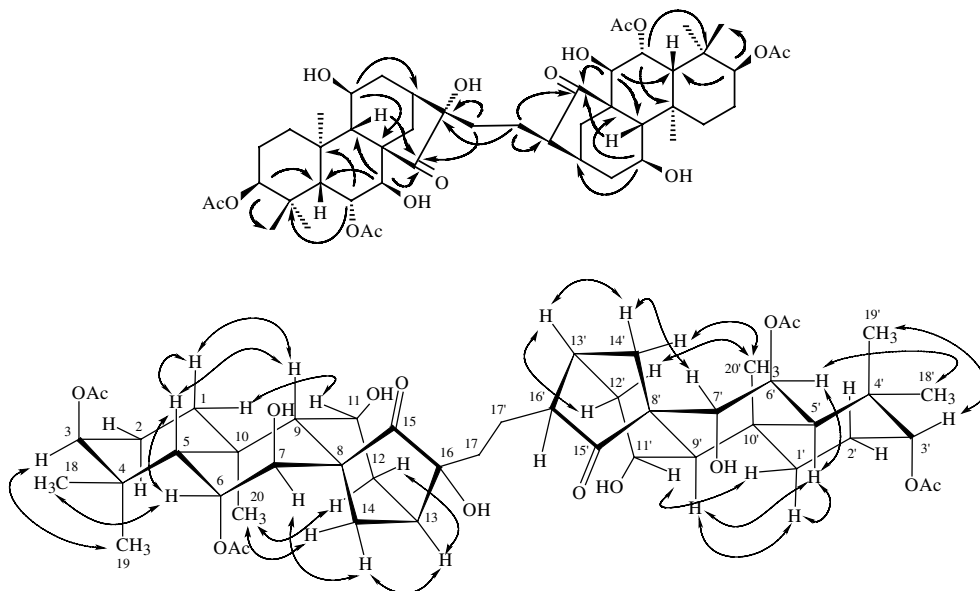


Figure 1. Selected HMBC and ROESY correlations of **1**.

(C-4, 8, and 10). Considering that all the kauranoids isolated from the genus *Isodon* possessed an *ent*-configuration, the two units (**1a** and **1b**) were presumed to also be *ent*-kauranes. A comparison of their NMR data with those of melissoidesin G (**4**, a known *ent*-kauranoid also isolated from this fraction) revealed that both **1a** and **1b** were very similar to **4** except at C-16 and C-17. The olefinic quaternary carbon (C-16) of **4** corresponded to an oxygenated quaternary carbon ( $\delta$  80.6) in **1a** and a methine carbon ( $\delta$  57.3) in **1b**, respectively, while the olefinic methylene carbon (C-17) of **4** was replaced by two methylene carbons ( $\delta$  30.7 and 20.8) in **1a** and **1b**. More importantly, the protons of these two methylene groups ( $\delta$  3.17 and 2.39;  $\delta$  2.97 and 2.66) showed  $^1\text{H}$ - $^1\text{H}$  COSY correlations with each other, indicating a single bond between these two carbons belonging to **1a** and **1b**. This was supported by a series of HMBC correlations between H<sub>2</sub>-17 with C-13, 15, 16, and 16' and H<sub>2</sub>-17' with C-13', 15', 16', and 16 (Fig. 1).

The substituents were accordingly assigned as 3 $\beta$ -OAc, 6 $\alpha$ -OAc, 7 $\beta$ -OH, 11 $\beta$ -OH, 3' $\beta$ -OAc, 6' $\alpha$ -OAc, 7' $\beta$ -OH, and 11' $\beta$ -OH, respectively, by the HMBC and ROESY correlations as shown in Figure 1. The OH-16 was suggested to be in an  $\alpha$ -orientation by the significant upfield signal of C-12 ( $\delta$  32.5) compared to that in **4**, which was caused by the  $\gamma$ -steric compression effect between 16 $\beta$ -methylene group with H-12 $\beta$ .<sup>1a</sup> Similarly, the  $\beta$ -orientation was deduced for 16'-methylene group. Finally, the structure of this diterpene dimer (**1**) was established as presented in Figure 1, which was confirmed by single crystal X-ray diffraction (Fig. 2).<sup>6</sup>

The positive HRFABMS and  $^{13}\text{C}$  NMR data (Table 1) revealed xindongnin N (**2**) to be an analogue of **1**, a diterpene dimer having the molecular formula C<sub>48</sub>H<sub>68</sub>O<sub>15</sub>.<sup>5</sup> It was suggested to possess a similar linkage between its two subunits, a single carbon-carbon bond

on the basis of the characteristic  $^{13}\text{C}$  NMR signals: an oxygenated quaternary carbon ( $\delta$  80.5, due to C-16), two methylene carbons ( $\delta$  30.5 and 20.5, C-17 and C-17') and a methine carbon ( $\delta$  57.1, C-16'). In contrast to **1**, xindongnin N (**2**) was constructed from two different diterpene units (**2a** and **2b**). **2a** is similar to **4**, while **2b** resembled xindongnin A (**5**). The assignment of the oxygenated quaternary carbon (or the methylene carbon at  $\delta$  30.5) was the key to the structural elucidation of **2**. Detailed analysis of the COSY, HMQC, and HMBC spectra of **2** disclosed that the protons ( $\delta$  3.05 and 2.26) of the methylene group ( $\delta$  30.5, C-17) exhibited HMBCs with both the oxygenated quaternary carbon (80.5) and a carbonyl carbon ( $\delta$  225.7). And this carbonyl carbon was further indicated to be C-15 of **2a** by the HMBC correlation between the characteristic H-7 ( $\delta$  4.31) of **2a** with this carbon. Therefore, the oxygenated quaternary carbon ( $\delta$  80.5) should be assigned to C-16 of **2a**, and the structure of **2** was established to be as shown. In the same way, xindongnin O (**3**) was elucidated to be an isomer of **2**, in whose structure the oxygenated quaternary carbon ( $\delta$  80.6) was indicated to be C-16 of the unit similar to xindongnin A (**5**). According to these dimers' structures, all the NMR data of **2** and **3** were successfully assigned with the aid of their 2D NMR spectra.

The biotransformation of this kind of dimers from normal *ent*-kauranoids isolated from the genus *Isodon*, was proposed to be as shown in Scheme 1. These *ent*-kauranoids normally had the  $\alpha,\beta$ -unsaturated ketone groups. The [4+2] cycloaddition, between the  $\alpha,\beta$ -unsaturated ketone group of one diterpene unit with the olefin group of a second unit, yielded a six-membered heterocycle,<sup>7</sup> which linked the monomers together. The hydrolysis and rearrangement at this heterocycle then produced the title dimers. We note that this [4+2] cycloaddition needs rigorous conditions in organic synthesis, which are impossible in plants. It is possible

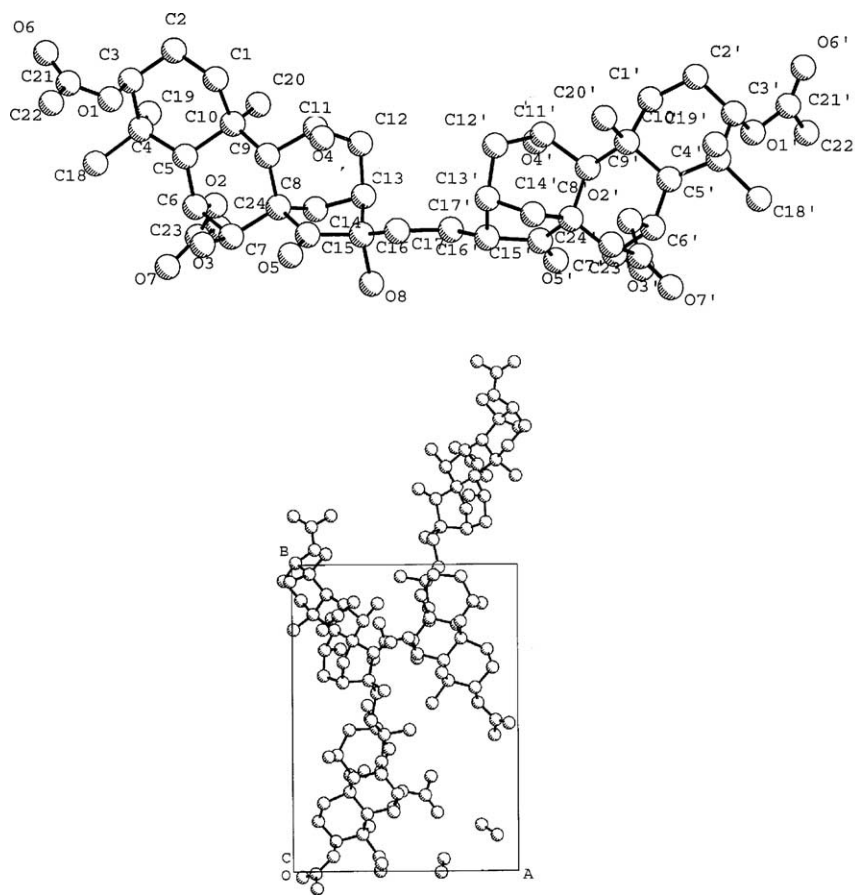
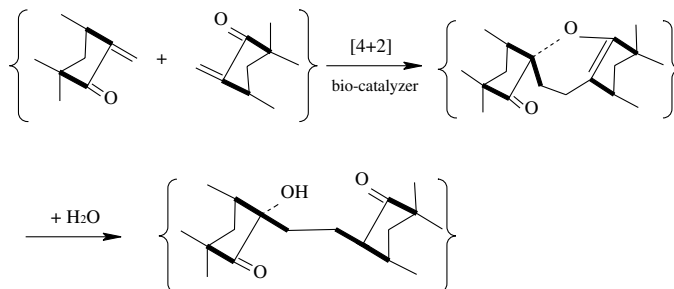


Figure 2. Crystal structure of compound 1.



Scheme 1. Proposed biogenesis of 1–3.

that in the *Isodon* plants there may be a bio-enzyme that catalyzes this reaction.

### References and notes

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- The leaves (1 kg) of *Isodon rubescens* var. *rubescens* were collected in Shangcheng Prefecture, Henan Province, PR China, in August 2002. The plant material was identified by Prof. Zhong-Wen Lin, and a voucher specimen (KIB-09-2002-Lin) was deposited in the Herbarium of the Department of Taxonomy, Kunming Institute of Botany, Chinese Academy of Sciences.
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- Compound 1: colorless cubes, mp 204–206 °C;  $[\alpha]_D^{22.0}$  –17.2 (MeOH, *c* 0.15); positive FABMS *m/z*: 887  $[M+H]^+$ ;

- positive HRFABMS  $m/z$ :  $[M+H]^+$  887.4811 (calcd for  $C_{48}H_{71}O_{15}$  887.4793); IR (KBr)  $\nu_{\max}$ : 3444, 2939, 2879, 1731, 1433, 1375, 1241, 1033  $cm^{-1}$ ;  $^1H$  NMR ( $C_5D_5N$ , 500 MHz) and  $^{13}C$  NMR ( $C_5D_5N$ , 125 MHz); see Table 1. Compound **2**: white amorphous powder;  $[\alpha]_D^{22.0}$   $-5.8$  (MeOH,  $c$  0.17); positive FABMS  $m/z$ : 885  $[M+H]^+$ ; positive HRFABMS  $m/z$ :  $[M+H]^+$  885.4633 (calcd for  $C_{48}H_{69}O_{15}$  885.4636); IR (KBr)  $\nu_{\max}$ : 3458, 2939, 1736, 1731, 1434, 1375, 1243, 1035  $cm^{-1}$ ;  $^1H$  NMR ( $C_5D_5N$ , 500 MHz) and  $^{13}C$  NMR ( $C_5D_5N$ , 125 MHz); see Table 1. Compound **3**: white amorphous powder;  $[\alpha]_D^{22.0}$   $-15.4$  (MeOH,  $c$  0.20) positive FABMS  $m/z$ : 885  $[M+H]^+$ ; positive HRFABMS  $m/z$ :  $[M+H]^+$  885.4635 (calcd for  $C_{48}H_{69}O_{15}$  885.4636); IR (KBr)  $\nu_{\max}$ : 3446, 2939, 1736, 1636, 1435, 1375, 1243, 1034  $cm^{-1}$ ;  $^1H$  NMR ( $C_5D_5N$ , 500 MHz) and  $^{13}C$  NMR ( $C_5D_5N$ , 125 MHz); see Table 1.
6. Crystal data for **1**. Crystals of **1**, crystallized from acetone, belong to the monoclinic space group  $P2_1$ . Crystal data:  $C_{48}H_{70}O_{15} \cdot (CH_3OH)_2 \cdot (H_2O)_{0.5}$ ,  $M = 887.07$ ,  $a = 12.826$  (1),  $b = 16.859$  (1),  $c = 12.872$  (1) Å,  $\beta = 104.56$  (1)°,  $V =$

2694.0(3) Å<sup>3</sup>,  $Z = 2$ ,  $d = 1.182$  g/cm<sup>3</sup>, Mo K $\alpha$  radiation, linear absorption coefficient  $\mu = 1.0$  cm<sup>-1</sup>. A colorless cube of dimensions 0.20×0.20×0.50 mm was used for X-ray measurements on a MAC DIP-2030K diffractometer with a graphite monochromator, maximum  $2\theta$  value of 50.0° was set. The total number of independent reflections measured was 4286, 4253 of which were considered to be observed ( $|F|^2 \geq 3\sigma|F|^2$ ). The structure was solved by the direct method SHELXS-86 and expanded using difference Fourier techniques, refined by the program and method NOMCSDP<sup>8</sup> and full-matrix least-squares calculations. Hydrogen atoms were fixed at calculated positions. The final indices were  $R_f = 0.084$ ,  $R_w = 0.086$  ( $w = 1/\sigma|F|^2$ ),  $S = 4.542$ ,  $(\Delta/\rho)_{\max} = 0.135$ ,  $(\Delta/\rho)_{\min} = -0.520$  e/Å<sup>3</sup>,  $(\Delta/\rho)_{\max} = 0.310$  e/Å<sup>3</sup>. The crystal structure of **1** has been deposited at the Cambridge Crystallographic Data Centre and allocated the deposition number CCDC 226868.

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